## d-d Band Excitation of the Diorganonickel( $\parallel$ ) Complexes *trans*-[Ni{2,6-(MeO)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>}<sub>2</sub>L<sub>2</sub>] (L = PMe<sub>3</sub> or PMe<sub>2</sub>Ph) and Their Analogues causing Reversible Phosphine Dissociation and/or Organic Group Rotation about the Ni–C Bond

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The title photo-reactions were observed to give stationary state mixtures, which are different from those resulting from thermal equilibrations.

Square-planar diorganonickel(II) complexes, trans-[NiR<sub>2</sub>L<sub>2</sub>], are commonly yellow to orange owing to the presence of the so-called 'd-d band' in their electronic spectrum.<sup>1</sup> The complexes (1a), (1b), (2a), (2b), (3a), and syn- and anti-(3b) also exhibit such a band<sup>†</sup> in the 360–400 nm region as a shoulder on very intense bands in the u.v. region tailing towards the visible region. Since the d-d excitation would transfer an electron in a non-bonding d orbital to the antibonding  $d_{x^2 - y^2}$  orbital, we thought that it would weaken the Ni–R and/or the Ni–L bonds. We have now studied the photoreaction of complexes (1)–(3) in order to investigate the excitation effects on their Ni–L and Ni–R bonds.

The reactions were monitored by the <sup>1</sup>H n.m.r. spectra of 0.03-0.05 M degassed solutions, as in our previous work on thermal reactions.<sup>2,3</sup> Complexes (1a) and (1b) are stable in benzene at 35 °C, and little change was observed in their <sup>1</sup>H n.m.r. spectra after 4 h irradiation by a 500 W Xe lamp. When a 1:1 mixture of (1a) and (1b) in benzene was irradiated for 1-8 h, the spectrum showed the formation of the mixed ligand complex (AB), *trans*-NiR<sub>2</sub>(PMe<sub>3</sub>)(PMe<sub>2</sub>Ph) (1ab) [equation (1)] together with a small amount of the decomposition product R<sub>2</sub> (Table 1). The proportion of (1ab) increased with irradiation time, although on prolonged irradiation the solution became turbid owing to decomposition. In the dark at 35 °C, much less (1ab) was formed. The use of a u.v. filter to remove light below 350 nm did not change the yield of (1ab) appreciably.

<sup>† (1</sup>a), 360; (1b), 375; (2a), 374; (2b), 397; anti-(3b), 385; syn-(3b), 384 nm;  $\varepsilon$  300–750 (in C<sub>6</sub>H<sub>6</sub> or CH<sub>2</sub>Cl<sub>2</sub>).

Con	nplex	Solvent	Time/h	(A):(B):(AB):(D)	
(A)	<b>(B)</b>			Irradiated	[Dark (24 h)]
(1a)	(Ìb)	C <sub>6</sub> H <sub>6</sub>	1	45:45:10:0	
		0 0	8	26:28:41:5	[47:49:4:0]
(2a)	( <b>2b</b> )	CH <sub>2</sub> Cl <sub>2</sub>	1	39:38:20:3	
` '			11	17:16:62:5	[46:43:11:0]

Table 1. Photo-induced intermolecular tertiary phosphine exchange<sup>a</sup> for complexes (1) and (2) [equation (1)].

<sup>a</sup> Degassed solutions of (A) and (B) (0.05 M each) were irradiated by a 500 W Xe lamp at 35 °C in n.m.r. tubes.



$$trans-[NiR_{2}L^{1}_{2}] + trans-[NiR_{2}L^{2}_{2}] \rightleftharpoons$$
(A)
(B)
$$2 trans-[NiR_{2}L^{1}L^{2}] + R-R \quad (1)$$

(AB) (D)

For complexes (2a) and (2b), the fifth and the sixth co-ordination sites of nickel are completely obstructed by the four methoxy groups because of the influence of the *meta*bromo substituents, as confirmed by an X-ray crystal structure determination,<sup>4</sup> and by the very low reactivity of these complexes towards carbon monoxide.<sup>2</sup> Photo-induced ligand exchange was still observed between these complexes, however, suggesting a mechanism involving three-coordinate intermediates.

A benzene solution containing (1a) and free  $PMe_2Ph$  (1.5 equiv.) gave on heating at 81 °C a thermal equilibrium mixture

of (1a), (1ab), and (1b) (78:22:1). Interestingly, on 4 h irradiation of this equilibrated mixture at 35 °C, the ratio changed to 56:41:3. On heating this solution, the thermal equilibrium was restored.

Complexes (3a) and (3b) exist in both *anti* and *syn* forms owing to the lack of free rotation of the organic group around the Ni–C bonds, and both isomers have been separated in pure form for (3b).<sup>3</sup> When a benzene solution of *anti*-(3b) was irradiated at 35 °C, it partially isomerized to *syn*-(3b). Analogous isomerization was observed for *syn*-(3b). Both isomerizations resulted in a stationary state with an *anti* to *syn* ratio of 68:32 after 8 h irradiation, a different ratio from the thermal equilibration (*anti*:*syn* 50:50).<sup>3</sup> In the dark at 35 °C, neither isomerization differences between the photo and thermal reactions were observed for (3a) to give a photostationary state with an *anti*:*syn* ratio of 64:36, compared with the thermally equilibrated ratio of 55:45.

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